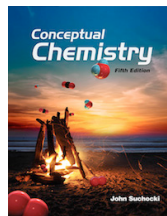


# Teacher's Manual

## Conceptual Chemistry — Life Science

2020 - 2021



Thank you for signing up for Conceptual Chemistry—the “Life Science” Version. You are now looking over the teacher’s manual (aka, user’s guide) to your Conceptual Academy course. There’s a lot to digest, but please understand: This course is truly a guided course. We authors are your tour guides. We know the path rather well having traveled it many times before. You, the teacher, are your student’s personal assistant. You are also their coach, cheering them onward and providing personalized help as needed. The purpose of this manual is to provide you, the teacher, a birds-eye view of an amazing adventure that’s about to unfold for both yourself and your student. But throughout your journey, for any further support you might need, please write to us at [Support@ConceptualAcademy.com](mailto:Support@ConceptualAcademy.com).

### 1. About Life Science Chemistry

This course is divided into 7 *units*, which are partitioned into a total of 30 *modules*. The amount of material covered is sufficient for a full academic year of study. Each module is designed to take one to two weeks to complete with the assumption the student has other courses and extracurricular activities. You’ll find this course focuses on material relevant to the life sciences serving as a good background for students with an interest in biology for allied health. Along with basic chemistry concepts, this includes biochemistry, nutrition, and pharmaceuticals. The content of this course is found primarily within the first 14 chapters of the *Conceptual Chemistry* textbook.

### 2. Laboratories

Science and experiments go together hand-in-glove. For this Conceptual Academy course you will find all the lab activities have been embedded. These labs are designed to be done with materials readily available in the home or a local discount store. We recommend the student keep a “field journal” to document their experience. Please look to the Doc Share of the first FYI page of the course for details regarding these labs. You’ll find instructor notes, grading rubrics and a material’s list.

You’ll find relatively few questions to be graded within the labs. For grading purposes, these few questions can be ignored. Their importance is minor. Instead, the full thrust of “grading” should be on the quality of the student’s field journal. Toward this you’ll find grading rubrics within the To the Student document. We place it there so that the student

can also see what's expected of them. While early entries might be sloppy and lacking, the goal is to see improvement as the year progresses. Periodically, expect that you'll need to buckle down on your field journal critiques.

In addition to the activities that you'll find throughout the course, you will also find the "Hands-On Chemistry" and "Confirm the Chemistry" activities found in the textbook. These are easy to perform activities using only household materials. For example, soak pennies in some salt water. This strips copper ions into solution. Place an iron nail in this solution and you will see the copper metal depositing on the nail within minutes. This activity is done in the context of electrochemistry as addressed in Chapter 11. There are about 4 or 5 of these short and sweet activities found within each chapter of the textbook. Again, the student's experience with these activities should be recorded in the field journal.

Also, there are about half a dozen interactive simulations already embedded into the course. For example, in Chapter 6, the student will encounter a simulation for building molecular models, which can be moved about in virtual 3D space. We see these valuable simulations as more minds-on than hands-on.

For students seeking a more traditional laboratory experience, we have teamed up with eScience Labs who offers three different Conceptual Academy customized chemistry kits we call: Helium, Neon, Argon. The Argon kit contains the most experiments and is the most expensive. This should be considered for cases where the student is clearly aiming for a career in a science-related field. A single Argon kit can also be nicely shared among several students who are taking the course together. In such a situation, we suggest each student takes a turn at being the "principal investigator". Their role is to direct the others on how to actually run the lab while staying "hands-off", which means this student has some prep work to do before everyone begins.

The Neon kit is a slimmed down (and more affordable) version of the Argon kit. The Helium kit, in turn, is a slimmed down version of the Neon kit and designed to maximize impact while minimizing cost. Still, any of these kits are relatively expensive. Of course, this is because it also includes equipment, which is shipped right to your home. For more details, see our support site: [LearnScience.Academy](http://LearnScience.Academy).

### **3. Learning Philosophy**

We are advocates of "Step 1/Step 2" learning. Step 1 is where the student is being introduced to material, such as through the textbook and video tutorials. Step 1 is an input process. You'll note the mouth is closed. "Step 2" is an output process where the student tries to articulate (output) that which they think they learned from Step 1 through activities such as presentations and homework. Of these two steps, Step 2 is arguably the more difficult. Students tend to avoid Step 2 or neglect its importance. Learning, however, is only deep and durable when BOTH Step 1 and Step 2 have been employed. See our "How to Study Effectively" document in the first Doc Share of this course to learn more.

#### **4. A Typical Student Day**

Please visit [HomeschoolPlanet.com](http://HomeschoolPlanet.com) where you will find detailed automated planners available for each Conceptual Academy course. These planners were built by the conceptual authors. Each planner details specific learning activities a student is to accomplish for each day across the entire course. That said, below are the general ideas that went behind the building of these planners.

What should a student's typical day look like? We find it generally best for the student to begin with a Step 1 activity, such as reading the textbook or watching a video tutorial. After completing each textbook chapter section or video, the student is encouraged to ask themselves a most powerful learning question: "What did I just learn?" and then to answer this very question aloud or in writing. In doing so, the student is interleaving a Step 2 activity within a Step 1 activity, which is most productive.

After working with the textbook and videos, the student should put effort into the "Practice Page" worksheets available from the Doc share within each module.

There are the many end-of-chapter (EOC) questions within the textbook, all of them partitioned by chapter section. These are an important Step 2 activity for the student. Because of their great number, we believe that half of the odd-numbered questions is fairly ambitious. Notably, the student will find the answers to the odd-numbered questions at the back of the book. This is important for the student to be able to confirm their understanding. It's also important that the student try their best to answer the question BEFORE looking at the answer. Any good answer will "make sense" after reading it. But it's not the answer that matters. What counts is being able to come up with the answer on one's own. A good student understands the world of difference between reading an answer and creating that answer him or herself.

Any opportunity the student has to summarize (aloud) what they believe they have learned to classmates, friends, or family members, is a serious bonus to the learning process—on many levels. You should consider the following capstone learning activity: Once a student "completes" a chapter, have the student provide a verbal summary of the main ideas of the chapter. You can call this: "The Summary Challenge". It's not as easy as it might sound, but it's a great way of identifying that which has been retained (or not). The process itself helps to make the learning durable. The student can then read the author's own chapter summary, which you'll find provided in the Doc Shares of the FYI pages. After reading the author's summary, ask the student to give their verbal summary another try. It's normal for students to stumble as they try their best to articulate what they think they learned. They deserve much credit for putting forth this kind of effort.

Then there are Conceptual Academy's "Homework Practice Sessions", also known as the HPS. The HPS serves a similar purpose to the end-of-chapter questions (Step 2 Learning). The goal is to provide the student ample opportunity to practice that which they think they have learned. As we'll describe shortly, the HPS questions are relatively difficult. Though there can be over 100 questions within a single HPS, the student is

expected to study only as many as they can in a single session. Working on 10 to 20 questions in a single sitting is respectable.

In addition to the above activities, there are the labs and the unit exams. Which is to say, there is no shortage of resources available for your student. Success can be had by shifting from one resource to the next. As soon as the student becomes saturated with one activity, then STOP. Move to another activity. Interleave these activities as per the wishes of the student. This is in contrast to a learning system of: “Hey kiddo. Here’s the textbook. Now read until you learn something. Then we’ll see if you can pass the test.” Not good! We know from experience that the key to helping students learn is offering a variety learning resources. Variety is YUM! Let your student’s days be filled with variety.

## **5. Grades—Summative and Formative**

In traditional academics, most students are more focused on their letter grade than the actual learning that letter grade is suppose to represent. When scholarships and admissions to competitive colleges are at stake, this is understandable. Ideally, though, a higher letter grade reflects higher learning. But there are all sorts of exceptions. A student who has struggled for a “B” in a subject that is of sincere interest, is more likely to retain that knowledge over the long term, than a book savvy student who could care less but can still pull an “A”.

We await the day when the standardized “Scholastic Aptitude Test” is replaced with a “Scholastic Attitude Test”. In our experience as college professors, attitude is just as important as aptitude, if not more so.

It’s not until graduate school that many students begin to realize that the whole A|B|C|D|F grade system itself is to be taken with a huge grain of salt. What counts most is the learning, which is closely related to good attitude. But more than mere “knowledge” the ideal goal is nurturing our innate curiosity. Then beyond curiosity, and much more valuable than a perfect SAT score, is helping the student grow into a responsible, well-adjusted, happy, loving, and productive individual who can support him or herself and a family with a career they actually enjoy.

The true value of a grade is not as a final end-all to a particular course of study. Let’s call that a “*summative grade*”. Rather, grades are more important as feedback that helps us learn DURING that course of study. Let’s call this a “*formative grade*”. The value in a formative grade is in the guidance it provides while we still have time to make corrections—to let us know when we’re on track and when we might still be holding onto misconceptions BEFORE the end of the semester.

We here at Conceptual Academy are not in a position to assign a final overall summative grade for each student taking one of our self-study courses. This is the responsibility of the teacher who has been working directly with the student throughout the course of study. We are, however, very much in a position to provide formative grades throughout.

This comes in the form of the reading quizzes, the video quizzes, and the HPS as tracked by the Conceptual Academy grade book, as well as the answers to the chapter questions at the back of the textbook, and the answers to the Practice Pages available within the Doc Shares, as well as the answer keys we provide for all unit exams and lab activities. Let's talk about these components one by one.

### *Reading Quiz*

A set of easy-to-answer questions collated from the chapter sections of each lesson. There is one "lesson reading quiz" for each lesson. These are designed to provide the student credit for having studied the paragraphs of the textbook. In many cases, the question is printed right within the margin of the textbook while the answer is highlighted within the paragraph. The "Reading Check" question at Conceptual Academy is the multiple choice version of that question.

### *Video Quiz*

A set of easy-to-answer questions relating to a particular video. Each video has a video quiz, which consists of one to three questions. These questions focus on the content of the video, which will be similar to the content of the corresponding textbook chapter section. In most cases, a student who has studied the textbook chapter section, may be able to do well with the video quiz without watching the video. This is by design to minimize any "busy work". We don't want the student watching the video if they don't need to watch the video. Similarly, the student might be able to do well on a lesson reading quiz without reading the textbook but after watching all the videos. Ideally, a student works with both the textbook and the videos, which together provide the student a stereoscopic view of the material.

### *Homework Practice Session (HPS)*

There is one Homework Practice Session (HPS) for each lesson. The HPS is an important Step 2 activity for the student after having studied the reading and video assignments for that lesson. The questions of the HPS are relatively difficult. Plus there are many of these questions—sometimes over 100 in a single session.

The goal of a homework practice session is PRACTICE. We liken it to practicing basketball. While practicing on the court, the basketball player should be taking chances, making mistakes, and learning from those mistakes. While practicing, the player should not be limited in the number of shots they're allowed to take. Also, the player should only practice for as long as is efficient. Push themselves, yes, but not to the point of exhaustion. When efficiency is lost, it's best to STOP. Take a break. Come back the next day to practice some more while fresh.

The same holds true for the Homework Practice sessions in which the student is encouraged to take chances, make mistakes, and learn from those mistakes. A session should go for as long as is reasonable for that student, which can vary from

day to day. For each correct answer, the student earns 2 points. There's never a penalty for any wrong answer.

Each question is first presented in a short answer format. The student's free response is not graded, nor recorded in any database. Rather, it serves as a warm up to the multiple choice version of that question, which comes next.

If a student is averaging about 55% on a set of HPS questions, then this is admirable. It shows they're trying to answer the question without first looking up the answer. This is to be encouraged. Students should also be encouraged to work with others on these questions, which makes for a good group activity.

### *Points from the Reading and Video Quizzes and the HPS*

For context, you should understand how the reading and video quizzes and the HPS are used at the college level. College students using Conceptual Academy are typically told they need to collect a certain number of CA points by the end of the semester. How many points depends upon the needs of the course, where 800 is a typical number. For such a course, all students who acquire at least 800 points will earn a 100% on this assignment, which counts for about 20% of their total summative letter grade. In this scenario, students aren't penalized for wrong answers. They just need to keep answering questions until they earn these 800 points. We call this an "encouragement-based approach" such that Conceptual Academy is there to reward students for good study habits.

So, in the college scenario, you've got one instructor with potentially 100+ students. For a self-study course where the student teacher ratio is closer to 1:1, then there's opportunity to take it to the next level. We suggest requiring your student to earn in the area of 1500 points by the end of the course, but paced along the way by having thresholds they must meet for each unit.

For this Conceptual Chemistry, Life Science, course, we recommend the following per unit grading scheme for the automated quizzes at Conceptual Academy:

#### **Per Unit Grading for the Automated Quizzes**

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

The student works their way up from one "metal status" to the next. Their goal is to reach the gold status for full credit "A" level work. Each subsequent level they might

reach (platinum and supernova) allows them to advance by one metal status in another unit where they might be shy in points earned.

The particular numbers in the above chart were derived from several considerations. First, the gold level is comparable to what diligent college students earn across the many college campuses using Conceptual Academy. Second, our experience with homeschools is that this 200 point unit threshold is an admirable level. Even more admirable is when the student maintains that gold level status throughout the entire course. Third, we have tabulated the number of actual points available for each unit for this course as follows:

All Units	Unit A	Unit B	Unit C	Unit D	Unit E	Unit F	Unit G
4546	838	900	854	382	658	540	374

There are well over 200 points available for each unit. This translates into ample opportunity for the student as well as a fair amount of flexibility. Remember, the goal with these automated quizzes is not to assess what students have learned, but to provide credit for their time on task learning.

In assigning a final summative letter grade, we recommend the student's Conceptual Academy score from the video, reading, and HPS questions (combined) should be worth from 20% to 40% of the overall course grade. The remaining portion of the overall course grade should be from the lab activities, unit exams, and any special projects.

### *End-of-Chapter Questions*

Most of the questions at the back of each textbook chapter are presented in the short answer format. But which ones should your student answer? Here's a good rule of thumb: Every other odd-numbered question is relatively ambitious. You'll note that the questions start out easy, then build in difficulty level. We feel the "Think and Explain" questions are of most value.

Now, "how" should your student answer these questions? Think of these questions as conversation starters. Ideally, the student has some one they can explain their thoughts to verbally. There's a discussion that leads to an agreement. Only then is the answer looked up in the back of the book. The student might rate themselves as to the quality of their initial answer on a scale of 1 through 5. At that point most students would just move onto the next question. We have a better alternative: Now that the student has been exposed to the "real answer", have them explain it again (without looking at the answer). Then they rate again on a scale of 1 through 5. You'll see what's happening here is the student is articulating. As this is done, there are pathways within the brain that are literally being built. Durable learning is occurring. It requires effort. No one is exempt. It can be tiring. What to do when the student gets exhausted and feels like a sponge with all the water squeezed out of it? Answer: Stop! Do something else. Some physical activity. Or study another subject, such as history. Or just take a break.

You should see that all the end-of-chapter questions are quite the resource for Step 2 learning. But how to grade their performance on these questions? The answer is: don't. Learning is still occurring. No grading please. But you might consider setting a goal for the number of questions worked upon. For example, 10 questions. Dear student: Get through 10 questions and you'll have earned yourself a sticker, or better yet, a chocolate bar. Ultimately, the student should recognize that working on these EOC questions is great preparation for the unit exam. But beyond that, learning is its own reward. If all students could be brought to this understanding, we believe the bulk of problems in our nation's education system would melt away.

### *Practice Page Worksheets*

These are pencil-pushing minds-on activities. In a way, they are similar to a lab experience, except it's all on paper. Our goal in creating these Practice Pages is to provide an enjoyable venue through which the student can *apply* what they think they understand. Please note: It's not like the student already understands something and then should be able to do the worksheet. It's the other way around. The understanding evolves only when the student is working on the worksheet.

This is similar to the end-of-chapter questions. The student may ask: How can I answer these questions if I don't first understand the material? They have it backward. The real question is: How can you understand the material if you don't first work on these questions? The understanding itself arises from working on the questions. After Step 1? After reading the book and watching the videos? The student may feel they still don't really understand. That's correct. That's a wise student! An even wiser student knows that the understanding will grow like a seed from the soil only when watered by a stream of well-placed questions (Step 2). We call this "formative".

### *Unit Exams*

Our unit exams can be considered a blend of both formative and summative grading, but with an emphasis on the formative. They are each presented in what at first seems like an unusual and complicated format, we call the "pyramid" format. But once you've been introduced to this format, you'll see it as a great learning opportunity. And fun too!

You'll find the pdf for each unit exam in the Doc Share on the last FYI page of each unit. Unlike the quizzes, and much like the Practice Pages, these are to be printed out. You'll find each exam begins with directions on how to run the exam in the pyramid format. Keep in mind that this format requires relatively tough level 3 questions. Your student is doing well upon earning around 60% on the first round. On subsequent rounds, this score will improve dramatically. It's important that students know of this design. Again, a 60% on round one is to be applauded.

In assigning a final summative grade, performance on these unit exams should hold some weight. For college students, their exams typically account from 50% to 80%



of their course grade (lecture component). But for this course, there are only six unit exams across the entire academic year. We suggest that the unit exam counts for around 20% of the overall grade for a particular unit. But we also suggest that you trust your own judgment for your own particular situation.

In our minds, if a student is doing all the work and has a good command over the material as evidenced by their ability to articulate that which they think they understand, then that qualifies as excellent. Attitude is everything.

### *Lab Activities*

Many colleges still follow a 3:1 credit system, whereby the student earns 3 credit hours for “lecture” and 1 credit hour for “laboratory”. On some campuses the student earns a single grade for both lab/lecture. At other campuses, the grade for lecture and lab are recorded separately. So, one way or the other, the lab component counts for about 25% of the overall assessment. It is typical that a student’s lab scores tend to lift their lecture scores, which are based primarily on the mid-term exams as described above.

For your Conceptual Academy course, we recommend the same kind of balance. Ideally, the hands-on “experiential” lab activities are there to complement the more minds-on process of learning concepts. The two work together. Relative to a grade, consider granting your student 50% just for completing a lab activity. You might then consider the remaining 50% on the quality of a student’s writing or their answers to questions as evidenced from their field journal. Toward this, we have provided lab activity grading rubrics (See the first Doc Share of the course)

### *Podcasts*

We are in the process of building a podcast series aligning to the chapters of the Conceptual Chemistry textbook and this course. No grading. But we expect many students will find much value in these podcast episodes, which are perfect for long trips, when folding laundry, cleaning one’s room, or the like. Perhaps there may be times when the student just needs a break or some inspiration. For that, please consider our podcast series. You’ll find we cover more than just the concepts of chemistry. You’ll also find interviews with well-known scientists and engineers and experts in education. We even offer audio field trips. As a bit of promotion, you’ll find we have embedded relevant episodes at key moments throughout the course.

On a final note, the course concludes with a downloadable, high-resolution Certificate of Completion, pre-signed by the author and awaiting the signature of the student’s mentor.

## 6. Putting It All Together

In addition to the textbook, our library of video lessons (integrated with the textbook), the automated quizzes, the HPS, the Practice Pages, the unit exams, you will also find study advice from the author on each FYI page, interactive simulations, plus a number of Easter egg surprises spiced here and there. And for technical support, please write to [Support@ConceptualAcademy.com](mailto:Support@ConceptualAcademy.com).

We know you will find this self-study course to be rich not only in content but in flavor. Our goals go beyond imparting knowledge. We aim to nurture a life-long curiosity about this amazing natural world in which we are blessed to live. We know this to be an important path to becoming good stewards. Further, the rules of nature are what we all have in common and as this world gets smaller, a focus on what we have in common becomes all the more important. Further still, understanding science for what it is, for what it can do, for what it can't do, for how it, for better or worse, has impacted our daily lives, is critically important for any student in this modern age.

Thank you for your support of Conceptual Academy. We are so please and honored to be working with you.

Good science to you!  
The Conceptual Academy Team



Prof. Hewitt  
Geology



Prof. Suchocki  
Chemistry & Astro



Prof. Hewitt  
Physics



Prof. Lyons  
Earth Science



Prof. Yeh  
Biology

## 7. But wait, there's more! The Log/Planner Sheets

What follows is the table of contents of this course that you might use as a log sheet to chart student progress and to demonstrate to any college admissions officer the college level quality of the chemistry course taken by your student.

To take planning to an even higher level, we have also created our planners now available at [HomeschoolPlanet.com](http://HomeschoolPlanet.com) fitting into their slick automated system. These planners were built by us conceptual authors. Each planner details specific learning activities a student is to accomplish for each day across the entire course.

## Syllabus: Conceptual Chemistry, Life Science

Log/Planner Sheet (Completion Date)

### Unit : A : Elements of Chemistry

#### Unit A Grading

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

#### Module 1 ( / / )

- FYI page
- 1.1 Science Is a Way of Understanding the Natural World
- 1.2 The Discovery of the Buckyball
- 1.3 Technology Is Applied Science
- 1.4 We Are Still Learning About the Natural World
- 1.5 Chemistry Is Integral to Our Lives
- 1.6 Scientists Measure Physical Quantities
- Module Reading Quiz
- Homework Practice Session

#### Module 2 ( / / )

- FYI page
- 2.1 The Submicroscopic World Is Super-Small
- 2.2 Discovering the Atom
- 2.3 Mass Is How Much and Volume Is How Spacious
- 2.4 Density Is the Ratio of Mass to Volume
- 2.5 Energy Is the Mover of Matter
- Module Reading Quiz
- Homework Practice Session

#### Module 3 ( / / )

- FYI page
- 2.6 Temperature Is a Measure of How Hot--Heat It Is Not
- 2.7 The Phase of a Material Depends on the Motion of Its Particles
- 2.8 Gas Laws Describe the Behavior of Gases
- Module Reading Quiz
- Homework Practice Session

#### Module 4 ( / / )

- FYI page
- 3.1 Matter Has Physical and Chemical Properties
- 3.2 Elements Are Made of Atoms

- 3.3 The Periodic Table Helps Us to Understand the Elements
- 3.4 Elements Can Combine to Form Compounds
- Module Reading Quiz
- Homework Practice Session

### Module 5 ( / / )

- FYI page
- 3.5 There Is A System for Naming Compounds
- 3.6 Most Materials Are Mixtures
- 3.7 Matter Can Be Classified as Pure or Impure
- 3.8 The Advent of Nanotechnology
- Module Reading Quiz
- Homework Practice Session

## Unit : B : Atomic Bonding

### Unit B Grading

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

### Module 1 ( / / )

- FYI page
- 4.1 Physical and Conceptual Models
- 4.2 The Electron Was the First Subatomic Particle Discovered
- 4.3 The Mass of an Atom Is Concentrated in its Nucleus
- 4.4 The Atomic Nucleus Is Made of Protons and Neutrons
- 4.5 Light Is a Form of Energy
- 4.6 Atomic Spectra and the Quantum Hypothesis
- Module Reading Quiz
- Homework Practice Session

### Module 2 ( / / )

- FYI page
- 5.1 Radioactivity--the Disintegration of the Atomic Nucleus
- 5.2 Radioactivity Is a Natural Phenomenon
- 5.3 Radioactivity Results from an Imbalance of Forces
- 5.4 Radioactive Elements Transmute to Different Elements
- 5.5 The Shorter the Half-Life, the Greater the Radioactivity
- 5.6 Isotopic Dating Measures the Ages of Materials
- Module Reading Quiz
- Homework Practice Session

**Module 3 ( / / )**

- FYI page
- 6.1 Electron-Dot Structures
- 6.2 Atoms Can Lose or Gain Electrons to Become Ions
- 6.3 Ionic Bonds Result from a Transfer of Electrons
- 6.4 The Electrons of Metallic Bonds Are Loosely Held
- Module Reading Quiz
- Homework Practice Session
- 

**Module 4 ( / / )**

- FYI page
- 6.5 Covalent Bonds Result from a Sharing of Electrons
- 6.6 Valence Electrons Determine Molecular Shape
- 6.7 Polar Covalent Bonds--Uneven Sharing of Electrons
- Module Reading Quiz
- Homework Practice Session

**Unit : C : Chemistry of Solutions****Unit C Grading**

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

**Module 1 ( / / )**

- FYI page
- 6.8 Molecular Polarity--Uneven Distribution of Electrons
- 7.1 Four Different Types of Dipole Attractions
- Module Reading Quiz
- Homework Practice Session

**Module 2 ( / / )**

- FYI page
- 7.2 A Solution Is a Single-Phase Homogeneous Mixture
- 7.3 Concentration Is Given As Moles Per Liter
- 7.4 Solubility Is How Well a Solute Dissolves
- 7.5 Soap Works by Being Both Polar and Nonpolar
- 7.6 Softening Hard Water
- 7.7 Purifying the Water We Drink
- Module Reading Quiz
- Homework Practice Session

### Module 3 ( / / )

- FYI page
- 8.1 Water Molecules Form an Open Crystalline Structure in Ice
- 8.2 Freezing and Melting Go On at the Same Time
- 8.3 Liquid Water's Behavior Results from the Stickiness of Its Molecules
- Module Reading Quiz
- Homework Practice Session

### Module 4 ( / / )

- FYI page
- 8.4 Water Molecules Move Freely Between the Liquid and Gaseous Phases
- 8.5 It Takes a Lot of Energy to Change the Temperature of Liquid Water
- 8.6 A Phase Change Requires the Input or Output of Energy
- Module Reading Quiz
- Homework Practice Session

## Unit : D : Chemical Reactions

### Unit D Grading

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

### Module 1 ( / / )

- FYI page
- 9.1 Chemical Reactions Are Represented by Chemical Equations
- 9.2 Counting Atoms and Molecules By Mass
- Module Reading Quiz
- Homework Practice Session

### Module 2 ( / / )

- FYI page
- 9.3 Converting Between Grams and Moles
- Module Reading Quiz
- Homework Practice Session

### Module 3 ( / / )

- FYI page
- 9.4 Chemical Reactions Can Be Exothermic or Endothermic
- 9.5 Chemical Reactions Are Driven By Entropy
- Module Reading Quiz
- Homework Practice Session

**Module 4 ( / / )**

- FYI page
- 9.6 Chemical Reactions Can Be Slow or Fast
- 9.7 Catalysts Speed Up the Destruction of Stratospheric Ozone
- Module Reading Quiz
- Homework Practice Session

**Unit : E : Acids, Bases, and Redox****Unit E Grading**

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

**Module 1 ( / / )**

- FYI page
- 10.1 Acids Donate Protons and Bases Accept Them
- 10.2 Some Acids and Bases Are Stronger Than Others
- 10.3 Solutions Can Be Acidic, Basic, or Neutral
- Module Reading Quiz
- Homework Practice Session
- 

**Module 2 ( / / )**

- FYI page
- 10.4 Buffer Solutions Resist Changes in pH
- 10.5 Rainwater Is Acidic
- 10.6 Carbon Dioxide Acidifies the Oceans
- Module Reading Quiz
- Homework Practice Session

**Module 3 ( / / )**

- FYI page
- 11.1 Losing and Gaining Electrons
- 11.2 Harnessing the Energy of Flowing Electrons
- 11.3 Batteries Consume Chemicals to Generate Electricity
- 11.4 Fuel Cells Consume Fuel to Generate Electricity
- Module Reading Quiz
- Homework Practice Session

**Module 4 ( / / )**

- FYI page
- 11.6 Electrolysis Produces Chemical Change
- 11.8 Oxygen Is Responsible for Corrosion and Combustion
- Module Reading Quiz
- Homework Practice Session

**Unit : F : Organic and Biochemistry****Unit F Grading**

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

**Module 1 ( / / )**

- FYI page
- 12.1 Hydrocarbons Contain Only Carbon and Hydrogen
- 12.2 Unsaturated Hydrocarbons Have Multiple Bonds
- 12.3 Functional Groups Give Organic Compounds Character
- Module Reading Quiz
- Homework Practice Session

**Module 2 ( / / )**

- FYI page
- 12.4 Alcohols, Phenols, and Ethers Contain Oxygen
- 12.5 Amines and Alkaloids Contain Nitrogen
- 12.6 Carbonyl Containing Compounds
- Module Reading Quiz
- Homework Practice Session

**Module 3 ( / / )**

- FYI page
- 12.7 An Example of Organic Synthesis
- 12.8 Organic Molecules Can Link to Form Polymers
- Module Reading Quiz
- Homework Practice Session

**Module 4 ( / / )**

- FYI page
- 13.1 Biomolecules Are Produced and Utilized by Organisms
- 13.2 Carbohydrates Give Structure and Energy
- 13.3 Lipids Are Insoluble in Water
- Module Reading Quiz
- Homework Practice Session



**Module 5 ( / / )**

- FYI page
- 13.4 Proteins Are Polymers of Amino Acids
- 13.5 Nucleic Acids Code for Proteins
- Module Reading Quiz
- Homework Practice Session

**Unit : G : Food and Drugs****Unit G Grading**

<i>Metal Status</i>	<i>Points to Qualify</i>	<i>Points Earned</i>	<i>Grade</i>
Supernova	300		Advance by another metal in another unit
Platinum	250		Advance by one metal in another unit
<b>Gold</b>	<b>200</b>		"A" work
Silver	150		"B" work
Bronze	100		"C" work
Copper	50		"D" work
Aluminum	25		Developing

**Module 1 ( / / )**

- FYI page
- 13.6 Vitamins Are Organic, Minerals Are Inorganic
- 13.7 Metabolism Is the Cycling of Biomolecules Through the Body
- 13.8 The Food Pyramid Summarizes a Healthful Diet
- 15.1 Humans Eat at All Trophic Levels
- Module Reading Quiz
- Homework Practice Session

**Module 2 ( / / )**

- FYI page
- 14.1 Medicines Are Drugs That Benefit the Body
- 14.2 The Lock-and-Key Model Guides the Synthesis New Medicines
- 14.3 Chemotherapy Cures the Host by Killing the Disease
- Module Reading Quiz
- Homework Practice Session

**Module 3 ( / / )**

- FYI page
- 14.4 The Nervous System Is a Network of Neurons
- 14.5 Psychoactive Drugs Alter the Mind or Behavior
- Module Reading Quiz
- Homework Practice Session

**Module 4 ( / / )**

- FYI page
- 14.6 Pain Relievers Inhibit the Transmission or Perception of Pain
- 14.7 Medicines for the Heart
- Module Reading Quiz
- Homework Practice Session